## Effect of A-Strain on a Synthesis of *cis*-Fused 4a-Aryloctahydro-1*H*-cyclopenta[*c*]pyridine Derivatives through Tandem Radical Cyclisation of an $\alpha$ -Acylamino–Polyene System

Shinzo Kano,\* Yoko Yuasa, Tsutomu Yokomatsu, Kenji Asami, and Shiroshi Shibuya

Tokyo College of Pharmacy, 1432-1 Horinouchi, Hachioji, Tokyo 192-03, Japan

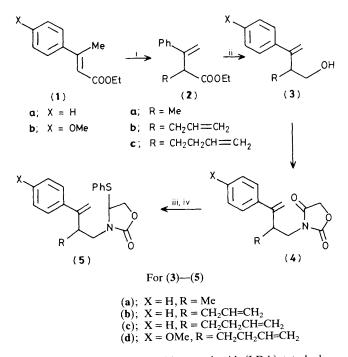
An efficient synthesis of the *cis*-fused 4a-aryloctahydro-1*H*-cyclopenta[*c*]pyridine ring system, an analogue of 4a-aryldecahydroisoquinoline, was achieved through a tandem radical approach by cyclisation of free radical–polyene species.

The construction of bicyclic systems via free radical cyclisation at an unsaturated component is becoming important.<sup>1</sup> In these reactions,  $\alpha$ -acylamino radicals have been used as versatile synthetic entries into N-heterocycles.<sup>2</sup> Recently, tandem radical cyclisation by the use of free radical-polyolefinic systems was applied to the synthesis of condensed carbocyclic systems.<sup>3</sup> Our interest in a polyene cyclisation strategy<sup>4</sup> for synthesis of potentially biologically active aza-polycyclic compounds led us to investigate a synthetic approach to the 4a-aryloctahydro-1*H*-cyclopenta[*c*]pyridine ring system. This is of interest from both synthetic and pharmacological points of view, since it is an analogue of 4a-aryldecahydroisoquinolines, which are significant as simple versions of morphine-based molecules.<sup>5</sup> Our strategy is based on a tandem radical cyclisation using polyene- $\alpha$ -amino radical species. The results of our studies are described herein. 4-Phenylthio-oxazolidin-2-ones (5a-d), used for the generation of radical species, were prepared as outlined in Scheme 1. Reduction of esters (2a—c), obtained by  $\alpha$ -alkylation of (1a), with LiAlH<sub>4</sub> gave the corresponding alcohols (3a-c),†

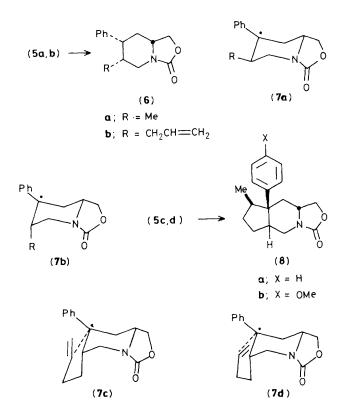
respectively. Condensation of  $(3\mathbf{a}-\mathbf{c})$  and  $(3\mathbf{d})^{4c}$  derived from (1b) with oxazolidine-2,4-dione by Mitsunobu's method<sup>6</sup> gave (4**a**-**d**), in 80-83% yield, respectively. Reduction of (4**a**-**d**), followed by phenylsulphenylation by modification of Walker's method<sup>7</sup> afforded (5**a**-**d**) in 70-75% yield, respectively.

A benzene solution of (5a) (0.01 M) was heated under reflux in the presence of tri-n-butyltin hydride (1.5 equiv.) and a trace amount of azobisisobutyronitrile (AIBN) in the usual way<sup>1,2</sup> to give (6a) in 57% yield as a single diastereoisomer, m.p. 124-128°C, m/z 231 ( $M^+$ ). The determination of the stereochemistry of (6a) was based on the Dreiding model study and Karplus relation<sup>8</sup> of signals due to NCH<sub>2</sub> and PhCH in its <sup>1</sup>H n.m.r. spectrum (CDCl<sub>3</sub>, 400 MHz). Of NCH<sub>2</sub>, the lower NCH signal appeared at  $\delta$  3.79 (d, J 13.36 Hz) and the higher one at  $\delta$  3.24 (dd, J 2.88, 13.36 Hz). PhCH resonated at  $\delta$  3.03 (broad d, J 12.88 Hz) owing to the large diaxial and small axial-equatorial interaction. These facts indicate that Me is axially and Ph is equatorially oriented. The Me signals, which appeared at considerably high field,  $\delta$  0.76 (d, J 7.00 Hz) also support this assignment. Diastereoselectivity in this cyclisation can be accounted for by the effect of A-type strain<sup>9</sup> on the benzyl radical intermediates. Of the two possible intermediates (7a, b; R = Me), (7b) should be

<sup>&</sup>lt;sup>+</sup> All new compounds gave satisfactory microanalyses and/or spectral data.



Scheme 1. Reagents: i, Lithium di-isopropylamide (LDA), tetrahydrofuran (THF), MeI[allyl bromide for (2b) and 1-iodobut-3-ene for (2c)], -78 °C—room temperature, 2 h; ii, LiAlH<sub>4</sub>, Et<sub>2</sub>O, 0 °C, 1 h; iii, NaBH<sub>4</sub>, MeOH, 0 °C; iv, PhSSPh, Bu<sup>n</sup><sub>3</sub>P, benzene, room temperature.



preferable to (7a) because of the steric repulsion of Me and Ph groups in (7a) in which Me is an equatorial substituent. The successive C-H bond formation via delivery of H to the less hindered face of the radical gave (6a). The same reaction using (5b) gave (6b) as a single diastereoisomer, yield 73%, m.p. 101-104 °C, m/z 257 (M<sup>+</sup>). However, in the case of (5c), further cyclisation of the intermediate (7c) occurred to form (8a) in 57% yield, m.p. 128–131 °C, m/z (M<sup>+</sup>), <sup>1</sup>H n.m.r. (CDCl<sub>3</sub>, 400 MHz) & 4.41 (1H, dd, J 8.48, 8.48 Hz), 3.97 (1H, dd, J 5.60, 8.48 Hz), 3.79 (1H, dd, J 1.48, 13.12 Hz), 3.01 (1H, dd, J 3.84, 13.12 Hz), 0.77 (3H, d, J 6.64 Hz). The ringjuncture of the 4a-phenylcyclopenta[c]pyridine ring was confirmed by the magnitude of the J value and the Karplus relation for the CH<sub>2</sub> signals at  $\delta$  3.79 and 3.01. Although the relative configuration of 5-Me was not determined at this stage, we assumed it to be *cis* to the Ph group from the intermediates (7c, d) for the second cyclisation. Of (7c, d), (7c) is more favourable than (7d) giving the *trans*-isomer. The considerably high field Me signal may support this assumption. In a similar way, (8b) was obtained from (5d) in 65% yield, m.p. 160—163 °C, m/z 301 ( $M^+$ ), <sup>1</sup>H n.m.r. (CDCl<sub>3</sub>, 400 MHz) δ 4.40 (1H, dd, J 8.40, 8.40 Hz), 3.96 (1H, dd, J 5.72, 8.40 Hz), 3.81 (3H, s), 3.67 (1H, dd, J 1.32, 13.60 Hz), 3.01 (1H, dd, J 3.84, 13.60 Hz), 0.76 (3H, d, J 6.72 Hz).

The method described in this paper should be widely applicable to the synthesis of condensed aza-polycyclic compounds.

Received, 1st August 1986; Com. 1107

## References

- G. Stork and N. H. Baine, J. Am. Chem. Soc., 1982, 104, 2321; G. Stork and R. Mook, Jr., *ibid.*, 1983, 105, 3720; G. Stork, R. Mook, Jr., S. A. Billar, and S. D. Rychonovsky, *ibid.*, p. 3741; M. D. Bachi, F. Frolow, and C. Hoornaert, J. Org. Chem., 1983, 48, 1841, and references cited therein.
- 2 D. J. Hart and Y.-M. Tasai, J. Am. Chem. Soc., 1982, 104, 1430; ibid., 1984, 106, 8201; ibid., p. 8209, and references cited therein.
- 3 D. P. Curran and D. M. Rakiewicz, J. Am. Chem. Soc., 1985, 107, 1448; D. P. Curran and M.-H. Chen, Tetrahedron Lett., 1985, 26, 4991.
- 4 (a) S. Kano, T. Yokomatsu, Y. Yuasa, and S. Shibuya, J. Org. Chem., 1985, 50, 3449: (b) S. Kano, T. Yokomatsu, H. Nemoto, and S. Shibuya, Chem. Lett, 1986, 143; (c) S. Kano, T. Yokomatsu, H. Nemoto, and S. Shibuya, J. Org. Chem., 1986, 51, 561.
- 5 M. R. Johnson and G. M. Milne, in 'Burger's Medicinal Chemistry,' 4th edn., ed. M. E. Wolf, Wiley Interscience, New York, 1981, Part III, pp. 699ff; D. C. Palmer and M. J. Straus, *Chem. Rev.*, 1977, 77, 1; W. H. Moos, R. D. Gless, and H. Rapoport, J. Org. Chem., 1981, 46, 5064, and references cited therein.
- 6 O. Mitsunobu, M. Wada, and T. Sano, J. Am. Chem. Soc., 1972, 94, 679.
- 7 K. A. M. Walker, Tetrahedron Lett., 1977, 4475.
- 8 L. M. Jackman and S. Sternhell, 'Application of Nuclear Magnetic Resonance Spectroscopy in Organic Chemistry,' 2nd edn., Pergamon Press, Oxford, 1969, Ch. 4-2, pp. 280ff.
- 9 F. Johnson, Chem. Rev., 1969, 68, 375.